



Editorial

International Journal of Clinical Pharmacology & Toxicology (IJCPT) ISSN 2167-910X

Toxic Compounds Analysis by High Performance Liquid Chromatography

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Received: September 17, 2014 Published: October 02, 2014

Citation: Hideharu Shintani (2014) Toxic Compounds Analysis By High Performance Liquid Chromatography. Int J Clin Pharmacol Toxicol. 3(3), 1-2. **doi:** http://dx.doi.org/10.19070/2167-910X-140005e

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I have studied the characteristics of toxic compounds from the sterilized medical devices or dental material, which migrated into patient's body fluid and evaluated if the migrated amount is risky or not. I have studied several sorts of toxic compounds. They are 4,4'-methylenedianiline (MDA), bisphenol A, phthalate esters, methylmethacrylate (MMA), N,N'-dimethyl p-toluidine (DMPT), benzoly peroxide (BPO), 4,4'-dihydroxydiphenyl sulfone, 4-hydroxy 4'-chloro diphenyl sulfone and so on. Mostly they are from medical devices and dental materials. Sorts of sterilization are gamma-ray irradiation, moist-heating, ethylene oxide (EOG) exposure. Detection of MDA and bisphenol A was by electro chemical detector (ECD) at 900 mV. BY applying 900 mV, MDA change to diimino compound and bisphenol A changes to keto compound. These amine-imino or keto-enol changeable tautomerization at relatively lower oxidation-reduction potential is essential factor to detect selectively by ECD.

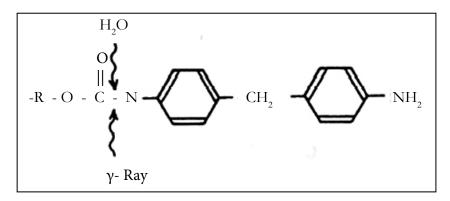
MDA can be produced from the gamma-ray sterilized thermoset-

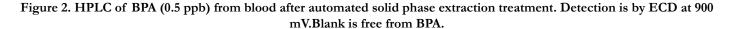
ting polyurethane (PU) and was found about 0.6 ppm for gammaray, 0.3 ppm for moist heat and 0.10 ppm for EOG, respectively [1]. This amount is from average three specimens. The virtual safe dose (VSD) was mathematically calculated by the one-hit model. Based on the VSD of 10⁻⁶ risk, estimation of the increased cancer risk from the highest MDA level of 0.6 ppm indicated a 1x10⁻⁷ increase for 50 kg patient exposed to dialysis three times per week for 70 years. This indicates virtually no serious risk. ECD detection for MDA from gamma-ray sterilized PU is the first report in the world. The formation mechanism of MDA by sterilization is that cleavage or hydrolysis at the urethane linkage adjacent to the terminal amino group by sterilization (Figure 1) [1]. The formation mechanism of MDA by sterilization is not simultaneous cleavage of consecutive urethane linkage, but only one cleavage at the urethane linkage adjacent to the terminal amino group as shown in Figure 1 [1].

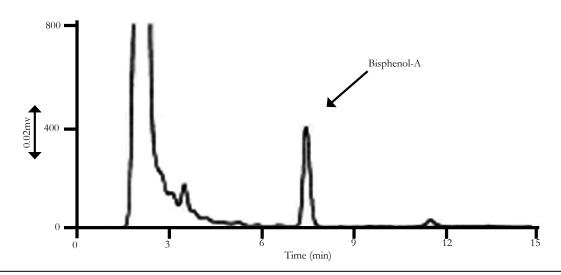
Second is the story about bisphenol A. Bisphenol A (BPA) is known carcinogen and endocrine disrupter. BPA is used to prepare polysulfone (PS) or polycarbonate (OC). PS is used for hollow fiber of dialyzer and PC is for housing of dialyzer. By gammaray or moist heat (121.1° C for 20 min) sterilization, bisphenol A is not detected by gamma-ray and 0.5 ppb was detected by moist heat [2]. This is quite low level detection, so no serious risk.HPLC chromatogram detected by ECD is presented in Figure 2.

Third is the phthalate ester compound. These are diethylhexyl phthalate (DEHP), monoethylhexyl phthalate (MEHP) and phthalic acid (PA). These are eluted from PVC blood bags into plasma and the migrated amount of DEHP, MEHP and PA to plasma was 270 ppm, 40 ppm and 0.6 ppm, respectively [3]. PA is non toxic, but DEHP and MEHP is seriously toxic and MEHP is more toxic than DEHP, which was first reported by the author. Their toxicity is serious, therefore must avoid to use DEHP in PVC. Currently safer plasticizer in place of DEHP is under study and preparation.

Figure 1. Proposed mechanism of MDA formation by cleavage. The arrows indicate the site of possible cleavage.







Next is the elution amount of MMA, DMPT, BPO to saliva. MMA, DMPT and BPO are the raw material of the preparation of methyl methacrylate polymer. MMA, DMPT and BPO are eluted into saliva at the amount of 32 ppm and 66 ppm, respectively. BPO immediately change to benzoic acid (BA), so BPO is hard to determine as BPO by HPLC. BPO can be determined as BA. The determined amount of BA was 2.2 ppm. Cytotoxicity data of IC50 (μ g/mL) of BPO, BA, DMPT and MMA using Balb 3T3 cell were 22, 28, 1500 and 4400, respectively [4]. BPO is most cytotoxic, but it cannot determine cytotoxicity as is because cell culture medium contains blood serum, so BPO immediately changes to BA, which information was obtained by the author at the first, BA is difficult to consider serious cytotoxic, but IC 50 data is shown to be seriously cytotoxic. This is hard to understand and must speculate the fact [4].

The last is the 4, 4'-dihydroxydiphenyl sulfone (bisphenol S) and 4-chloro 4'-hydroxy diphenyl sulfone. They are the raw material of polysulfone and they also present toxicity. Toxicity can be evaluated from the inhibition degree of microorganisms. Sterilization is by moist-heat (121.1° C for 15 min) or ozone gas sterilization (3000 ppm at 90 % RH). Microorganisms are *Geobacillus stearothermophilus* ATCC 12980 as a representative spore and *Micrococcus luteus*. The sterilization result was cultivated in SCDA medium at 55°C for *G. stearothermophilus* ATCC 12980 and SCDA medium at 30°C for *M.luteus* and was scoured colony forming unit (CFU) in SCDA plate. *G. stearothermophilus* ATCC 12980 was around half sterilized and *M.luteus*. was totally sterilized by both sterillization methods. The produced amount of 4-chloro 4'-hydroxy diphenyl sulfone, Bisphenol S and Bisphenol A was 625 ppm, 355 ppm and 43 ppm. As the amount of bisphenol A was less than others, so bisphenol A may not be a major factor for inhibitors [5].

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